

# Andhra Pradesh State Council of Higher Education

## Notations :

- 1.Options shown in green color and with ✓ icon are correct.
- 2.Options shown in red color and with ✗ icon are incorrect.

<b>Question Paper Name :</b>	Metallurgical Engineering 19th Sep 2021 Shift2
<b>Duration :</b>	180
<b>Total Marks :</b>	200
<b>Display Marks:</b>	No
<b>Calculator :</b>	None
<b>Magnifying Glass Required? :</b>	No
<b>Ruler Required? :</b>	No
<b>Eraser Required? :</b>	No
<b>Scratch Pad Required? :</b>	No
<b>Rough Sketch/Notepad Required? :</b>	No
<b>Protractor Required? :</b>	No
<b>Show Watermark on Console? :</b>	Yes
<b>Highlighter :</b>	No
<b>Auto Save on Console? ( SA type of questions will be always auto saved ) :</b>	Yes
<b>Is this Group for Examiner? :</b>	No

Section Id :	477203417
Section Number :	1
Mandatory or Optional :	Mandatory
Number of Questions :	50
Section Marks :	50
Enable Mark as Answered Mark for Review and Clear Response :	Yes

Question Number : 1 Question Id : 47720321233 Display Question Number : Yes Is Question Mandatory : No

If  $k \neq -5$  is a real number, then, the number of solutions to the following system of equations

$$3x - y + 4z = 3$$

$$x + 2y - 3z = -2$$

$$6x + 5y + kz = -3 \quad \text{is}$$

Options :

1. ✘ 0

2. ✔ 1

3. ✘ 2

4. ✘ *infinitely many*

Question Number : 2 Question Id : 47720321234 Display Question Number : Yes Is Question Mandatory : No

$$\begin{vmatrix} 1 & 1+p & 1+p+q \\ 2 & 3+2p & 4+3p+2q \\ 3 & 6+3p & 10+6p+3q \end{vmatrix} =$$

Options :

1. ✘ 0

2. ✔ 1

3. ✘ 2

4. ✘ 3

Question Number : 3 Question Id : 47720321235 Display Question Number : Yes Is Question

Mandatory : No

Let  $|A|$  denote the determinant of the matrix  $A$ . If  $A$  is a square matrix of order 3, and  $|4A| = r|A|$ , then the value of  $r$  is

Options :

1. ✘ 0

2. ✘ 4

3. ✘ 16

4. ✔ 64

Question Number : 4 Question Id : 47720321236 Display Question Number : Yes Is Question

Mandatory : No

If  $\begin{vmatrix} y & y \\ 1 & y \end{vmatrix} = \begin{vmatrix} 3 & 4 \\ 1 & 2 \end{vmatrix}$ , then the value of  $y$  is

Options :

1. ✘ 0

2. ✘ 1

3. ✔ 2

4. ✘ 3

Question Number : 5 Question Id : 47720321237 Display Question Number : Yes Is Question Mandatory : No

Let  $\begin{vmatrix} 2 & 3+i & -1 \\ 3-i & 0 & -1+i \\ -1 & -1-i & 1 \end{vmatrix} = a + ib$ , where  $a$  and  $b$  are real numbers. Then the value of  $b$  is

Options :

1. ✔ 0

2. ✘ 1

3. ✘ 3

4. ✘ 4

Question Number : 6 Question Id : 47720321238 Display Question Number : Yes Is Question Mandatory : No

If  $\frac{y^2-5y+1}{(y+1)(y+2)(y+3)} = \frac{a}{y+1} + \frac{b}{(y+1)(y+2)} + \frac{c}{(y+1)(y+2)(y+3)}$ , then,

Options :

1. ✘  $a = 1, b = 10, c = 25$
2. ✔  $a = 1, b = -10, c = 25$
3. ✘  $a = 5, b = 10, c = 25$
4. ✘  $a = 5, b = -10, c = 25$

Question Number : 7 Question Id : 47720321239 Display Question Number : Yes Is Question Mandatory : No

$$\frac{2x+3}{(x^2+1)(x+4)} =$$

Options :

1. ✘  $\frac{5}{17(x+4)} + \frac{5x+14}{17(x^2+1)}$
2. ✘  $\frac{-5}{17(x+4)} - \frac{5x+14}{17(x^2+1)}$
3. ✔  $\frac{-5}{17(x+4)} + \frac{5x+14}{17(x^2+1)}$

$$\frac{-5}{17(x+4)} + \frac{5x-14}{17(x^2+1)}$$

4. ✘

**Question Number : 8 Question Id : 47720321240 Display Question Number : Yes Is Question Mandatory : No**

If  $x$  and  $y$  are two distinct real numbers, then, the number of values of  $\theta$  in  $[0, 2\pi]$  for which  $\operatorname{cosec} \theta = \frac{x^2 - y^2}{x^2 + y^2}$  is

**Options :**

1. ✔ 0

2. ✘ 1

3. ✘ 2

4. ✘ 3

**Question Number : 9 Question Id : 47720321241 Display Question Number : Yes Is Question Mandatory : No**

If  $\cos(\alpha - \beta) + \cos(\beta - \gamma) + \cos(\gamma - \alpha) = -\frac{3}{2}$ , then  $\cos \alpha + \cos \beta + \cos \gamma =$

**Options :**

1. ✘  $-\frac{3}{2}$

2. ✘  $-1$

3. ✔ 0

4. ✘ 1

**Question Number : 10 Question Id : 47720321242 Display Question Number : Yes Is Question Mandatory : No**

For all real numbers  $\theta$ , the value of  $\sin^2 \theta + \cos^4 \theta$  is greater than or equal to

**Options :**

1. ✔  $\frac{3}{4}$

2. ✘ 1

3. ✘  $\frac{5}{4}$

4. ✘ 2

**Question Number : 11 Question Id : 47720321243 Display Question Number : Yes Is Question Mandatory : No**

Let  $x$  be a real number such that  $\tan\left(\frac{\pi}{4} + x\right) + \tan\left(\frac{\pi}{4} - x\right) = 2$ . Then  $x$  is of the form  $x = n\pi + a$ , where  $n \in \mathbb{Z}$ , and  $a =$

**Options :**

1. ✔ 0

2. ✘  $\frac{\pi}{3}$

3. ✘  $\frac{\pi}{4}$

4. ✘  $\frac{\pi}{6}$

Question Number : 12 Question Id : 47720321244 Display Question Number : Yes Is Question Mandatory : No

If  $(\sin^{-1} x) > (\cos^{-1} x)$ , then  $x$  belongs to the interval

Options :

1. ✘  $[0, \frac{1}{\sqrt{2}})$

2. ✔  $(\frac{1}{\sqrt{2}}, 1]$

3. ✘  $[\frac{1}{\sqrt{2}}, 1]$

4. ✘  $[0, \frac{1}{\sqrt{2}}]$

Question Number : 13 Question Id : 47720321245 Display Question Number : Yes Is Question Mandatory : No

Consider a triangle  $\Delta ABC$ , with sides of length  $a, b$  and  $c$ , and angles  $A, B$  and  $C$ . If

$a, b, c$  and the area of the triangle  $\Delta ABC$  are all rational, then

Options :

1. ✘  $\tan \frac{B}{2}$  is rational and  $\tan \frac{C}{2}$  is irrational.

2. ✘



$\tan \frac{B}{2}$  is irrational and  $\tan \frac{C}{2}$  is rational.

3. ✓  $\tan \frac{B}{2}$  and  $\tan \frac{C}{2}$  are both rational.

4. ✗  $\tan \frac{B}{2}$  and  $\tan \frac{C}{2}$  are both irrational.

**Question Number : 14 Question Id : 47720321246 Display Question Number : Yes Is Question Mandatory : No**

Consider a triangle  $\triangle ABC$ , with sides of length  $a, b$  and  $c$ , and angles  $A, B$  and  $C$ . If

$3a=b+c$ , then the value of  $\cot \frac{B}{2} \cdot \cot \frac{C}{2}$  is

**Options :**

1. ✗ 0

2. ✗  $\frac{1}{2}$

3. ✗  $\sqrt{3}$

4. ✓ 2

**Question Number : 15 Question Id : 47720321247 Display Question Number : Yes Is Question Mandatory : No**

$2 \tan^{-1} \left( \frac{3}{4} \right) - \tan^{-1} \left( \frac{17}{31} \right) =$

**Options :**

1. ✘ 0

2. ✔  $\frac{\pi}{4}$

3. ✘  $\frac{\pi}{2}$

4. ✘  $\pi$

**Question Number : 16 Question Id : 47720321248 Display Question Number : Yes Is Question Mandatory : No**

Consider a triangle  $\triangle ABC$  with angles  $A, B$  and  $C$ . If  $\cos A + \cos B + \cos C = \frac{3}{2}$ , then the triangle  $\triangle ABC$  is

**Options :**

1. ✔ equilateral.

2. ✘ isosceles, and right-angled.

3. ✘ isosceles, with one of the angles equal to  $\frac{\pi}{6}$ .

4. ✘ scalene

**Question Number : 17 Question Id : 47720321249 Display Question Number : Yes Is Question Mandatory : No**

The value of  $\cos^2 x + \cos^2 \left(x + \frac{\pi}{3}\right) + \cos^2 \left(x - \frac{\pi}{3}\right)$  is

Options :

1. ✘ 1

2. ✔  $\frac{3}{2}$

3. ✘ 2

4. ✘  $\frac{5}{2}$

Question Number : 18 Question Id : 47720321250 Display Question Number : Yes Is Question Mandatory : No

The value of  $\left(\frac{\sqrt{3}+i}{\sqrt{3}-i}\right)^3$  is

Options :

1. ✘  $-2\sqrt{2}$

2. ✔  $-1$

3. ✘ 0

4. ✘  $2\sqrt{2}$

Question Number : 19 Question Id : 47720321251 Display Question Number : Yes Is Question Mandatory : No

If  $x + iy = \frac{a+ib}{a-ib}$ , then  $x^2 + y^2 =$

Options :

1. ✘ 0

2. ✔ 1

3. ✘ 2

4. ✘ 4

Question Number : 20 Question Id : 47720321252 Display Question Number : Yes Is Question Mandatory : No

If a circle of radius 5 touches the circle  $x^2 + y^2 - 2x - 4y = 20$  at the point  $(5,5)$ , then, its center is

Options :

1. ✘ (8,8)

2. ✘ (8,9)

3. ✔ (9,8)

4. ✘ (9,9)

Question Number : 21 Question Id : 47720321253 Display Question Number : Yes Is Question Mandatory : No

The equation  $9x^2 - 24xy + 16y^2 - 20x - 15y = 60$  represents

Options :

1. ✓ a parabola

2. ✘ an ellipse

3. ✘ a hyperbola

4. ✘ a circle

**Question Number : 22 Question Id : 47720321254 Display Question Number : Yes Is Question Mandatory : No**

Let  $(x_j, y_j), j=1,2,3,4$ , be points of intersection of the parabola  $y^2 = 4ax$  and the circle  $x^2 + y^2 + 2gx + 2fy + c = 0$ .

Then  $y_1 + y_2 + y_3 + y_4 =$

**Options :**

1. ✘  $-2$

2. ✘  $-\frac{1}{2}$

3. ✓  $0$

4. ✘  $\frac{1}{2}$

**Question Number : 23 Question Id : 47720321255 Display Question Number : Yes Is Question Mandatory : No**

The length of the major axis of the ellipse  $9x^2 + 5y^2 - 30y = 0$  is

Options :

1. ✘  $\sqrt{5}$

2. ✘ 3

3. ✘  $2\sqrt{5}$

4. ✔ 6

Question Number : 24 Question Id : 47720321256 Display Question Number : Yes Is Question Mandatory : No

If  $S(-1, 1)$  is one of the foci of a hyperbola,  $x - y + 3 = 0$  is its directrix corresponding to  $S$  and 3 is its eccentricity, then, the equation of the hyperbola is

Options :

1. ✘  $7x^2 + 18xy + 7y^2 + 50x + 50y + 77 = 0$

2. ✘  $7x^2 + 18xy + 7y^2 + 50x - 50y + 77 = 0$

3. ✔  $7x^2 - 18xy + 7y^2 + 50x - 50y + 77 = 0$

4. ✘  $7x^2 - 18xy - 7y^2 - 50x + 50y + 77 = 0$

Question Number : 25 Question Id : 47720321257 Display Question Number : Yes Is Question Mandatory : No

The equation  $4(x - 2y + 1)^2 + 9(2x + y + 2)^2 = 25$  represents

**Options :**

1. ✘ a parabola
2. ✔ an ellipse
3. ✘ a hyperbola
4. ✘ a circle

**Question Number : 26 Question Id : 47720321258 Display Question Number : Yes Is Question**

**Mandatory : No**

Let  $f$  be a twice differentiable function such that  $f''(x) + f(x) = 0$ , and  $f'(x) = g(x)$ . If  $h(x) = [f(x)]^2 + [g(x)]^2$ , and  $h(10) = 20$ , then  $h(40) =$

**Options :**

1. ✔ 20
2. ✘ 40
3. ✘ 80
4. ✘ 160

**Question Number : 27 Question Id : 47720321259 Display Question Number : Yes Is Question**

**Mandatory : No**

$$\lim_{x \rightarrow \frac{\pi}{2}} \left( \frac{\cot x - \cos x}{\cos^2 x} \right) =$$

Options :

1. ✘  $-1$

2. ✔  $0$

3. ✘  $\sqrt{3}$

4. ✘  $\frac{\pi}{2}$

Question Number : 28 Question Id : 47720321260 Display Question Number : Yes Is Question Mandatory : No

Let  $\mathbb{R}$  be the set of all real numbers. Let  $f: \mathbb{R} \rightarrow \mathbb{R}$  satisfy the condition:

$|f(x) - f(y)| \leq |x - y|^{2021}$ , for all  $x, y \in \mathbb{R}$ . Then the value of  $f'(2022)$  is

Options :

1. ✔  $0$

2. ✘  $1$

3. ✘  $2021$

4. ✘  $2022$



Question Number : 29 Question Id : 47720321261 Display Question Number : Yes Is Question



**Mandatory : No**

The number of real roots of the equation  $x + e^x = 0$  is

**Options :**

1. ✘ 0

2. ✔ 1

3. ✘ 2

4. ✘ Infinitely many

**Question Number : 30 Question Id : 47720321262 Display Question Number : Yes Is Question**

**Mandatory : No**

If  $y = \text{Tan}^{-1} \left( \frac{\sqrt{1+\sin x} + \sqrt{1-\sin x}}{\sqrt{1+\sin x} - \sqrt{1-\sin x}} \right)$ , then  $\frac{dy}{dx} =$

**Options :**

1. ✘  $\cot^2 x$

2. ✘  $\sec^2 x$

3. ✔  $-\frac{1}{2}$

4. ✘  $\frac{1}{2}$

**Question Number : 31 Question Id : 47720321263 Display Question Number : Yes Is Question Mandatory : No**

The equation of the tangent to the curve  $x = \sin 3t, y = \cos 2t$ , at  $t = \frac{\pi}{4}$  is given by

**Options :**

1. ✘  $\sqrt{2}x - 3y - 2 = 0$

2. ✘  $\sqrt{2}x + 3y - 2 = 0$

3. ✔  $2\sqrt{2}x - 3y - 2 = 0$

4. ✘  $2\sqrt{2}x - 3y + 2 = 0$

**Question Number : 32 Question Id : 47720321264 Display Question Number : Yes Is Question Mandatory : No**

An open tank with a square base (with side  $x$ ) and vertical sides (with height  $y$ ) is to be constructed from a metal sheet so as to hold a given quantity of water. The cost of the material will be the least if

**Options :**

1. ✘  $x=y$

2. ✔  $x=2y$

3. ✘  $2x=y$

4. ✘

$$4x=y$$

**Question Number : 33 Question Id : 47720321265 Display Question Number : Yes Is Question Mandatory : No**

The function  $f(x) = x^3 - 12x^2 + 36x + 48$ , is decreasing in the interval

**Options :**

1. ✘  $(-\infty, 2)$

2. ✘  $(-\infty, 6)$

3. ✔  $(2, 6)$

4. ✘  $(6, \infty)$

**Question Number : 34 Question Id : 47720321266 Display Question Number : Yes Is Question Mandatory : No**

A shopkeeper can buy  $x$  items for Rs.  $\left(\frac{x}{5} + 500\right)$ . He can sell the  $x$  items at the rate

Rs.  $\left(5 - \frac{x}{100}\right)$  per item. Then the number of items he should sell to make maximum profit is

**Options :**

1. ✔ 240

2. ✘ 360

3. ✘ 400

4. ✘ 500

Question Number : 35 Question Id : 47720321267 Display Question Number : Yes Is Question Mandatory : No

$$\text{If } z = ax^2 + 2hxy + by^2, \text{ then } x \frac{\partial z}{\partial x} + y \frac{\partial z}{\partial y} =$$

Options :

1. ✘  $z$

2. ✘  $z^2$

3. ✘  $\frac{1}{2}z$

4. ✔  $2z$

Question Number : 36 Question Id : 47720321268 Display Question Number : Yes Is Question Mandatory : No

$$\int_{-1}^1 \frac{x \sin^{-1} x}{\sqrt{1-x^2}} dx =$$

Options :

1. ✘ 0

2. ✘ 1

3. ✘  $\frac{3}{2}$

4. ✔ 2

**Question Number : 37 Question Id : 47720321269 Display Question Number : Yes Is Question Mandatory : No**

The area of the region bounded by the curve  $y = x^2 + 4$ , the x-axis and the ordinates at  $x=1$  and  $x=5$  is

**Options :**

1. ✘  $\frac{147}{3}$

2. ✔  $\frac{172}{3}$

3. ✘  $\frac{187}{3}$

4. ✘  $\frac{227}{3}$

**Question Number : 38 Question Id : 47720321270 Display Question Number : Yes Is Question Mandatory : No**

$$\lim_{n \rightarrow \infty} \sum_{k=0}^{n-1} \frac{1}{\sqrt{n^2 - k^2}} =$$

**Options :**

1. ✘ 0

2. ✔

$$\frac{\pi}{2}$$

3. ✘  $\pi$

4. ✘  $2\pi$

Question Number : 39 Question Id : 47720321271 Display Question Number : Yes Is Question Mandatory : No

$$\int_0^1 \frac{2x}{1+x^2} dx =$$

Options :

1. ✘ 1

2. ✘ 2

3. ✔  $\log 2$

4. ✘  $3 \log 2$

Question Number : 40 Question Id : 47720321272 Display Question Number : Yes Is Question Mandatory : No

$$\int \frac{e^{ax} - e^{-ax}}{e^{ax} + e^{-ax}} dx =$$

(In the following,  $c$  is a constant.)

Options :

1. ✓  $\frac{1}{a} \log |e^{ax} + e^{-ax}| + c$

2. ✗  $\frac{1}{a} \log |e^{ax} - e^{-ax}| + c$

3. ✗  $\frac{1}{2a} \log |e^{ax} + e^{-ax}| + c$

4. ✗  $\frac{1}{2a} \log |e^{ax} - e^{-ax}| + c$

Question Number : 41 Question Id : 47720321273 Display Question Number : Yes Is Question Mandatory : No

$$\int_0^{\pi} \frac{e^{\cos x}}{e^{\cos x} + e^{-\cos x}} dx =$$

Options :

1. ✗  $-\pi$

2. ✗  $0$

3. ✓  $\frac{\pi}{2}$

4. ✗  $\pi$

Question Number : 42 Question Id : 47720321274 Display Question Number : Yes Is Question Mandatory : No

$$\int_{-\pi}^{\pi} \sin^5 x \, dx =$$

Options :

1. ✓ 0

2. ✗  $\frac{\pi}{2}$

3. ✗  $\pi$

4. ✗  $2\pi$

Question Number : 43 Question Id : 47720321275 Display Question Number : Yes Is Question Mandatory : No

The area of the region bounded by  $y=|x+3|$ , the x-axis and the lines  $x = -6$  and  $x = 0$  is

Options :

1. ✗ 3 square units

2. ✓ 9 square units

3. ✗ 12 square units

4. ✗ 18 square units

Question Number : 44 Question Id : 47720321276 Display Question Number : Yes Is Question



**Mandatory : No**

The degree of the differential equation  $7x \left(\frac{dy}{dx}\right)^2 - \frac{d^2y}{dx^2} + 10y = \log x$  is

**Options :**

1. ✓ 1

2. ✗ 2

3. ✗ 3

4. ✗ 4

**Question Number : 45 Question Id : 47720321277 Display Question Number : Yes Is Question**

**Mandatory : No**

The solution of the differential equation  $\frac{dy}{dx} = y \tan x$ , given that  $y=1$  when  $x=0$ , is given by

**Options :**

1. ✗  $y = \cos x$

2. ✗  $y = \cos 2x$

3. ✓  $y = \sec x$

4. ✗  $y = \sec 2x$

Question Number : 46 Question Id : 47720321278 Display Question Number : Yes Is Question

Mandatory : No

The solution to the differential equation  $(3x^2 + y) \frac{dx}{dy} = x$ , ( $x > 0$ ), such that  $y=1$  if  $x=1$  is

Options :

1. ✘  $y = 2x^2 - x$

2. ✔  $y = 3x^2 - 2x$

3. ✘  $y = 4x^2 - 3x$

4. ✘  $y = 5x^2 - 4x$

Question Number : 47 Question Id : 47720321279 Display Question Number : Yes Is Question

Mandatory : No

The differential equation of the family of parabolas having vertex at the origin and axis along the positive y-axis is

Options :

1. ✘  $xy' = 2$

2. ✔  $xy' = 2y$

3. ✘  $xy' = -2y$

4. ✘  $xy' = 2y^2$



Question Number : 48 Question Id : 47720321280 Display Question Number : Yes Is Question

**Mandatory : No**

The solution of the differential equation  $\frac{dy}{dx} + y \cot x = 4x \operatorname{cosec} x$ , ( $x \neq 0$ ), given that  $y=0$  when  $x = \frac{\pi}{2}$  is

**Options :**

$$y \operatorname{cosec} x = x^2 - \frac{\pi^2}{4}$$

1. ✘

$$y \operatorname{cosec} x = 2x^2 - \frac{\pi^2}{2}$$

2. ✘

$$y \sin x = x^2 - \frac{\pi^2}{4}$$

3. ✘

$$y \sin x = 2x^2 - \frac{\pi^2}{2}$$

4. ✔

**Question Number : 49 Question Id : 47720321281 Display Question Number : Yes Is Question**

**Mandatory : No**

The general solution of the differential equation  $\log_e \left( \frac{dy}{dx} \right) = ax + by$  is given by

**Options :**

$$ae^{ax} + be^{-by} + C = 0$$

1. ✘

$$ae^{ax} - be^{-by} + C = 0$$

2. ✘

$$\frac{1}{a}e^{ax} + \frac{1}{b}e^{-by} + C = 0$$

3. ✔

4. ✘  $\frac{1}{a}e^{ax} - \frac{1}{b}e^{-by} + C = 0$

Question Number : 50 Question Id : 47720321282 Display Question Number : Yes Is Question Mandatory : No

The particular integral of the differential equation  $(D^2 + D - 2)y = \sin x$  is given by

Options :

1. ✘  $-\frac{1}{10}(\cos x + \sin x)$

2. ✔  $-\frac{1}{10}(\cos x + 3 \sin x)$

3. ✘  $-\frac{1}{10}(\cos 3x + \sin 3x)$

4. ✘  $-\frac{1}{10}(3 \cos x + \sin x)$

## Physics

Section Id :	477203418
Section Number :	2
Mandatory or Optional :	Mandatory
Number of Questions :	25
Section Marks :	25
Enable Mark as Answered Mark for Review and	Yes

Clear Response :

Question Number : 51 Question Id : 47720321283 Display Question Number : Yes Is Question Mandatory : No

The dimensional formula for gravitational constant, G is

Options :

1. ✘  $M^1L^3T^{-2}$

2. ✔  $M^{-1}L^3T^{-2}$

3. ✘  $M^0L^3T^{-2}$

4. ✘  $M^2L^3T^{-2}$

Question Number : 52 Question Id : 47720321284 Display Question Number : Yes Is Question Mandatory : No

Which of the following quantities have not been expressed in proper units?

Options :

1. ✘ electric field = Newton/Coulomb

2. ✘ surface tension = Newton/meter

3. ✔ energy = kg m/s

4. ✘ pressure = Newton/m<sup>2</sup>

Question Number : 53 Question Id : 47720321285 Display Question Number : Yes Is Question

Mandatory : No

A vector A is along positive x-axis. If B is another vector such that  $A \times B$  is zero, then B could be

Options :

1. ✘  $4\hat{j}$

2. ✔  $-4\hat{i}$

3. ✘  $-(\hat{i} + \hat{j})$

4. ✘  $(\hat{j} + \hat{k})$

Question Number : 54 Question Id : 47720321286 Display Question Number : Yes Is Question

Mandatory : No

The scalar product of two vectors is  $2\sqrt{3}$  and the magnitude of their vector product is 2.  
The angle between them is

Options :

1. ✔  $30^\circ$

2. ✘  $45^\circ$

3. ✘  $60^\circ$

4. ✘  $90^\circ$



Question Number : 55 Question Id : 47720321287 Display Question Number : Yes Is Question

**Mandatory : No**

The work done by a force is defined as  $W = \mathbf{F} \cdot \mathbf{S}$ . In a certain situation  $\mathbf{F}$  and  $\mathbf{S}$  are not zero but the work done is zero when

**Options :**

1. ✘  $\mathbf{F}$  and  $\mathbf{S}$  are in the same direction
2. ✘  $\mathbf{F}$  and  $\mathbf{S}$  are in opposite direction
3. ✔  $\mathbf{F}$  and  $\mathbf{S}$  are at right angles
4. ✘  $\mathbf{F}$  and  $\mathbf{S}$  are at  $45^\circ$

**Question Number : 56 Question Id : 47720321288 Display Question Number : Yes Is Question**

**Mandatory : No**

A body starts from rest and travels a distance  $x$  in first two seconds and a distance  $y$  in next two seconds. The relation between  $x$  and  $y$  is

**Options :**

1. ✘  $y = 4x$
2. ✘  $y = x$
3. ✔  $y = 3x$
4. ✘  $y = 2x$

**Question Number : 57 Question Id : 47720321289 Display Question Number : Yes Is Question**

**Mandatory : No**

A projectile is projected with initial velocity  $(6\hat{i} + 8\hat{j})$  m/s. If  $g = 10 \text{ m/s}^2$  then horizontal range is

**Options :**

1. ✘ 4.8 m
2. ✔ 9.6 m
3. ✘ 19.2 m
4. ✘ 14.0 m

**Question Number : 58 Question Id : 47720321290 Display Question Number : Yes Is Question**

**Mandatory : No**

The maximum range of a projectile fired with some initial velocity is found to be 1000 m/s, in the absence of wind and air resistance. The maximum height reached by this projectile is

**Options :**

1. ✔ 250 m
2. ✘ 500 m
3. ✘ 1000 m
4. ✘ 2000 m

**Question Number : 59 Question Id : 47720321291 Display Question Number : Yes Is Question**



**Mandatory : No**

The force of friction between two bodies is

**Options :**

1. ✓ parallel to the contact surface
2. ✗ perpendicular to the contact surface
3. ✗ inclined at  $30^0$  to the contact surface
4. ✗ inclined at  $60^0$  to the contact surface

**Question Number : 60 Question Id : 47720321292 Display Question Number : Yes Is Question**

**Mandatory : No**

A body is sliding down an inclined plane under its own weight at constant speed. If the inclination of the plane to the horizontal is  $30^0$ , the angle of friction is

**Options :**

1. ✓  $30^0$
2. ✗  $60^0$
3. ✗  $45^0$
4. ✗  $90^0$

Question Number : 61 Question Id : 47720321293 Display Question Number : Yes Is Question

Mandatory : No

A block of mass 5 kg is resting on a smooth surface. At what angle, a force of 20 N be acted on the body so that it will acquire a kinetic energy of 40 J after moving 4m

Options :

1. ✘  $30^0$

2. ✘  $45^0$

3. ✔  $60^0$

4. ✘  $120^0$

Question Number : 62 Question Id : 47720321294 Display Question Number : Yes Is Question

Mandatory : No

Two men with the weights in the ratio 4:3 run up a staircase in time, in the ratio 12:11. The ratio of power of the first to that of second is

Options :

1. ✘  $\frac{4}{3}$

2. ✘  $\frac{12}{11}$

3. ✘  $\frac{48}{33}$

4. ✔  $\frac{11}{9}$

Question Number : 63 Question Id : 47720321295 Display Question Number : Yes Is Question Mandatory : No

Energy harnessed from flowing water is called-----energy

Options :

1. ✘ Solar
2. ✔ Hydel
3. ✘ Tidal
4. ✘ Geothermal

Question Number : 64 Question Id : 47720321296 Display Question Number : Yes Is Question Mandatory : No

The total mechanical energy of a spring-mass system in simple harmonic motion is  $E = 0.5 m\omega^2 A^2$ . If the oscillating particle is replaced by another particle of double the mass while the amplitude  $A$  remains the same. The new mechanical energy is

Options :

1. ✘  $2E$
2. ✘  $0.5 E$
3. ✘  $\sqrt{2} E$
4. ✔  $E$

Question Number : 65 Question Id : 47720321297 Display Question Number : Yes Is Question

Mandatory : No

Sound of frequency 1000 Hz from a stationary source is reflected from an object approaching the source at 30 m/s back to a stationary observer located at the source. The speed of sound in air is 330 m/s. The frequency of the sound heard by the observer is

Options :

1. ✓ 1200 Hz
2. ✗ 1000 Hz
3. ✗ 1090 Hz
4. ✗ 1100 Hz

Question Number : 66 Question Id : 47720321298 Display Question Number : Yes Is Question

Mandatory : No

The frequency of a pendulum if it is taken from the earth's surface to deep into a mine

Options :

1. ✗ increases
2. ✓ decreases
3. ✗ first increases then decreases
4. ✗ remains unchanged

**Question Number : 67 Question Id : 47720321299 Display Question Number : Yes Is Question Mandatory : No**

Two waves of lengths 50 cm and 51 cm produced 12 beats per second. The velocity of sound is

**Options :**

1. ✘ 340 m/s
2. ✘ 2. 331 m/s
3. ✔ 306 m/s
4. ✘ 360 m/s

**Question Number : 68 Question Id : 47720321300 Display Question Number : Yes Is Question Mandatory : No**

According to reverberation time the final intensity is around

**Options :**

1. ✘ one-hundredth of the initial intensity
2. ✘ one-tenth of the initial intensity
3. ✘ one-thousandth of the initial intensity
4. ✔ one-millionth of the initial intensity

Question Number : 69 Question Id : 47720321301 Display Question Number : Yes Is Question Mandatory : No

An ideal gas has volume  $V$  at pressure  $P$  and temperature  $T$ . Mass of each molecule is  $m$ . The density of the gas is

Options :

1. ✘  $mKT$

2. ✘  $\frac{P}{KT}$

3. ✘  $\frac{P}{KTV}$

4. ✔  $\frac{Pm}{KT}$

Question Number : 70 Question Id : 47720321302 Display Question Number : Yes Is Question Mandatory : No

Work done by 0.1 mole of a gas at  $27^{\circ}\text{C}$  to double its volume at constant pressure is  
( $R=2$  cal/mol/K)

Options :

1. ✘ 54 cal

2. ✘ 600 cal

3. ✔ 60 cal

4. ✘

546 cal

**Question Number : 71 Question Id : 47720321303 Display Question Number : Yes Is Question Mandatory : No**

If the pressure of a gas contained in a closed vessel is increased by 0.4%, when heated by  $1^{\circ}\text{C}$ , its initial temperature is

**Options :**

1. ✓ 250 K

2. ✗ 150 K

3. ✗ 100 K

4. ✗ 50 K

**Question Number : 72 Question Id : 47720321304 Display Question Number : Yes Is Question Mandatory : No**

A monoatomic ideal gas, initially at temperature  $T_1$  is enclosed in a cylinder fitted with a frictionless piston. The gas is allowed to expand adiabatically to a temperature  $T_2$  by releasing the piston suddenly. If  $L_1$  and  $L_2$  are the lengths of the gas column, before and after expansion respectively,  $T_1/T_2$  is given by

**Options :**

1. ✗  $\left(\frac{L_1}{L_2}\right)^{2/3}$

2. ✓  $\left(\frac{L_2}{L_1}\right)^{2/3}$

3. ✘  $\frac{L_1}{L_2}$

4. ✘  $\frac{L_2}{L_1}$

**Question Number : 73 Question Id : 47720321305 Display Question Number : Yes Is Question Mandatory : No**

A Carnot's engine operates with source at  $127^{\circ}\text{C}$  and sink at  $27^{\circ}\text{C}$ . If the source supplies 40 kJ of heat energy, the work done by the engine is

**Options :**

1. ✘ 30 kJ

2. ✔ 10 kJ

3. ✘ 4 kJ

4. ✘ 1 kJ

**Question Number : 74 Question Id : 47720321306 Display Question Number : Yes Is Question Mandatory : No**

The optical fibre consisting of a central core is clad by material of

**Options :**

1. ✔ slightly lower refractive index

2. ✘



slightly higher refractive index

equal refractive index

3. ✘

very high refractive index

4. ✘

Question Number : 75 Question Id : 47720321307 Display Question Number : Yes Is Question Mandatory : No

The susceptibility of the superconductor is

Options :

1. ✘ positive and small

2. ✘ negative and small

3. ✘ positive and unity

4. ✔ negative and unity

## Chemistry

Section Id :	477203419
Section Number :	3
Mandatory or Optional :	Mandatory
Number of Questions :	25
Section Marks :	25

Enable Mark as Answered Mark for Review and  
Clear Response :

Yes

Question Number : 76 Question Id : 47720321308 Display Question Number : Yes Is Question  
Mandatory : No

The nucleus of tritium consists of -----

Options :

1. ✘ 1 proton + 1 neutron
2. ✘ 1 proton + 3 neutrons
3. ✘ 1 proton + zero neutron
4. ✔ 1 proton + 2 neutrons

Question Number : 77 Question Id : 47720321309 Display Question Number : Yes Is Question  
Mandatory : No

Which of the following electronic configuration is not possible?

Options :

1. ✘  $1s^2 2s^2 2p^6$
2. ✔  $1s^2 2s^2 2p^7$
3. ✘  $1s^2 2s^2$
4. ✘  $1s^2 2s^2 2p^5$

Question Number : 78 Question Id : 47720321310 Display Question Number : Yes Is Question Mandatory : No

Radius of 3<sup>rd</sup> Bohr orbit of hydrogen atom is -----

Options :

1. ✘  $6.529A^0$

2. ✔  $4.761A^0$

3. ✘  $2.116A^0$

4. ✘  $8.464A^0$

Question Number : 79 Question Id : 47720321311 Display Question Number : Yes Is Question Mandatory : No

Covalent compounds are generally soluble in -----

Options :

1. ✔ Non-polar solvents

2. ✘ Polar solvents

3. ✘ Concentrated acids

4. ✘ All solvents

Question Number : 80 Question Id : 47720321312 Display Question Number : Yes Is Question Mandatory : No

Six electrons are mutually shared in -----

Options :

1. ✘ F<sub>2</sub>

2. ✘ Cl<sub>2</sub>

3. ✘ O<sub>2</sub>

4. ✔ N<sub>2</sub>

Question Number : 81 Question Id : 47720321313 Display Question Number : Yes Is Question Mandatory : No

To half the molarity of a solution, the following should be adopted.

Options :

1. ✘ Weight of the solute to be doubled

2. ✘ Weight of the solvent to be doubled

3. ✘ Volume of the solvent to be doubled

4. ✔ Volume of the solution to be doubled

**Question Number : 82 Question Id : 47720321314 Display Question Number : Yes Is Question Mandatory : No**

The molecular weight of  $\text{KMnO}_4$  is "M". In a reaction  $\text{KMnO}_4$  is reduced to  $\text{K}_2\text{MnO}_4$ . The equivalent weight of  $\text{KMnO}_4$  is

**Options :**

1. ✓ M
2. ✗  $M/2$
3. ✗  $M/3$
4. ✗  $M/4$

**Question Number : 83 Question Id : 47720321315 Display Question Number : Yes Is Question Mandatory : No**

Calculate the weight of NaOH present in 500 ml of 0.5 N Solution

**Options :**

1. ✗ 5 g
2. ✓ 10 g
3. ✗ 12 g
4. ✗ 15 g

**Question Number : 84 Question Id : 47720321316 Display Question Number : Yes Is Question Mandatory : No**

On addition of NaOH to water

Options :

1. ✘ Ionic product will increase
2. ✘ Ionic product will decrease
3. ✔ No change in ionic product of water
4. ✘  $\text{H}_3\text{O}^+$  concentration increases

Question Number : 85 Question Id : 47720321317 Display Question Number : Yes Is Question Mandatory : No

Which of the following is not a buffer solution?

Options :

1. ✘  $(\text{CH}_3\text{COOH}/\text{CH}_3\text{COONa})$
2. ✔  $(\text{HCl}/\text{NaCl})$
3. ✘  $(\text{HCOOH}/\text{HCOONa})$
4. ✘  $(\text{NH}_4\text{OH}/\text{NH}_4\text{Cl})$

Question Number : 86 Question Id : 47720321318 Display Question Number : Yes Is Question Mandatory : No

Which of the following is a good conductor of electricity?

Options :

1. ✘ Diamond
2. ✔ Graphite
3. ✘ Solid NaCl
4. ✘ Wood

Question Number : 87 Question Id : 47720321319 Display Question Number : Yes Is Question Mandatory : No

Which of the following (1M) conducts more electricity?

Options :

1. ✘ Acetic acid
2. ✘ Boric acid
3. ✘ Phosphorous acid
4. ✔ Sulphuric acid

Question Number : 88 Question Id : 47720321320 Display Question Number : Yes Is Question Mandatory : No

In electrolysis of dilute  $\text{H}_2\text{SO}_4$ , which of the following is liberated at anode in presence of inert electrode?

Options :

1. ✘ H<sub>2</sub>
2. ✘ SO<sub>2</sub>
3. ✔ O<sub>2</sub>
4. ✘ SO<sub>3</sub>

Question Number : 89 Question Id : 47720321321 Display Question Number : Yes Is Question Mandatory : No

The EMF of the cell Ni/Ni<sup>2+</sup> (0.01M)/ Cl<sup>-</sup>(0.01M)/Cl<sub>2</sub>, Pt is ---V if the SRP of nickel and chlorine electrodes are -0.25V and +1.36V respectively

Options :

1. ✘ + 1.61
2. ✘ - 1.61
3. ✔ + 1.79
4. ✘ - 1.79

Question Number : 90 Question Id : 47720321322 Display Question Number : Yes Is Question Mandatory : No

Which of the following is correct relation used to measures the hardness of water?



**Options :**

1. ✓  $1 \text{ mg/L} = 1 \text{ ppm} = 0.07^\circ\text{Cl} = 0.1^\circ\text{Fr}$
2. ✗  $1 \text{ mg/L} = 0.1 \text{ ppm} = 0.7^\circ\text{Cl} = 0.1^\circ\text{Fr}$
3. ✗  $1 \text{ mg/L} = 1 \text{ ppm} = 0.7^\circ\text{Cl} = 0.01^\circ\text{Fr}$
4. ✗  $1 \text{ mg/L} = 1 \text{ ppm} = 0.7^\circ\text{Cl} = 1^\circ\text{Fr}$

**Question Number : 91 Question Id : 47720321323 Display Question Number : Yes Is Question Mandatory : No**

Which of the following is used as effective coagulant in the municipal water treatment to remove fine suspended and colloidal impurities?

**Options :**

1. ✗  $\text{Fe}_2\text{SO}_4(\text{NH}_4)_2\text{SO}_4 \cdot 7\text{H}_2\text{O}$
2. ✓  $\text{K}_2\text{SO}_4 \cdot \text{Al}_2(\text{SO}_4)_3 \cdot 24\text{H}_2\text{O}$
3. ✗  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
4. ✗  $\text{Na}_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$

**Question Number : 92 Question Id : 47720321324 Display Question Number : Yes Is Question Mandatory : No**

The general chemical formula of zeolite is

Options :

1. ✓  $\text{Na}_2\text{O} \cdot \text{Al}_2\text{O}_3 \cdot x \text{SiO}_2 \cdot y \text{H}_2\text{O}$
2. ✗  $\text{Al}_2\text{O}_3 \cdot \text{H}_2\text{O}$
3. ✗  $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$
4. ✗  $\text{MgSO}_4 \cdot 5\text{H}_2\text{O}$

Question Number : 93 Question Id : 47720321325 Display Question Number : Yes Is Question Mandatory : No

----- is resulted when electrochemical corrosion happened in acidic environment.

Options :

1. ✗ Evolution of oxygen
2. ✗ Absorption of oxygen
3. ✓ Evolution of hydrogen
4. ✗ Absorption of hydrogen

Question Number : 94 Question Id : 47720321326 Display Question Number : Yes Is Question Mandatory : No

Impure metal corrodes faster than pure metal due to

**Options :**

1. ✓ Heterogeneity
2. ✗ Homogeneity
3. ✗ Non-galvanic cell
4. ✗ localize corrosion

**Question Number : 95 Question Id : 47720321327 Display Question Number : Yes Is Question Mandatory : No**

The number of repeating units in a polymer is called

**Options :**

1. ✗ Functionality
2. ✗ Tacticity
3. ✓ degree of polymerization
4. ✗ Specificity

**Question Number : 96 Question Id : 47720321328 Display Question Number : Yes Is Question Mandatory : No**

The process of vulcanisation makes rubber -----

**Options :**

1. ✘ Soft
2. ✔ Hard
3. ✘ Elastic
4. ✘ Swells oils

**Question Number : 97 Question Id : 47720321329 Display Question Number : Yes Is Question Mandatory : No**

Which of the following is thermosetting plastic

**Options :**

1. ✘ PVC
2. ✘ Polystyrene
3. ✘ Teflon
4. ✔ Bakelite

**Question Number : 98 Question Id : 47720321330 Display Question Number : Yes Is Question Mandatory : No**

The boiling range of petrol fraction is found to be

**Options :**

1. ✘  $120^{\circ}\text{C}-180^{\circ}\text{C}$

2. ✘  $250^{\circ}\text{C}-320^{\circ}\text{C}$

3. ✔  $40^{\circ}\text{C}-120^{\circ}\text{C}$

4. ✘  $180^{\circ}\text{C}-250^{\circ}\text{C}$

**Question Number : 99 Question Id : 47720321331 Display Question Number : Yes Is Question Mandatory : No**

Which of the following is not a common component of photochemical smog?

**Options :**

1. ✘ Ozone

2. ✘ Acrolein

3. ✘ Peroxyacetyl nitrate

4. ✔ Chlorofluorocarbons

**Question Number : 100 Question Id : 47720321332 Display Question Number : Yes Is Question Mandatory : No**

White lung cancer is caused by

**Options :**

1. ✘ Asbestos

2. ✔ Textiles

3. ✖ Paper

4. ✖ Silica

## Metallurgical Engineering

Section Id :	477203420
Section Number :	4
Mandatory or Optional :	Mandatory
Number of Questions :	100
Section Marks :	100
Enable Mark as Answered Mark for Review and Clear Response :	Yes

Question Number : 101 Question Id : 47720321333 Display Question Number : Yes Is Question

Mandatory : No

An ore is a naturally occurring aggregate or a combination of ....., from which one or more ..... or ..... may be extracted

Options :

1. ✖ Minerals, metals, compound

2. ✖ Metals, compounds, minerals

3. ✔ Minerals, metals, minerals

4. ✖ Metals, minerals, compounds

Question Number : 102 Question Id : 47720321334 Display Question Number : Yes Is Question Mandatory : No

Grizzly are commercial ore dressing units which work on the principle of:

Options :

1. ✘ Surface area
2. ✔ Size difference
3. ✘ Specific Gravity
4. ✘ Affinity for oxygen

Question Number : 103 Question Id : 47720321335 Display Question Number : Yes Is Question Mandatory : No

Sea-nodules are rich in:

Options :

1. ✘ Mn only
2. ✘ Ni only
3. ✔ Mn and Ni
4. ✘ Mg

Question Number : 104 Question Id : 47720321336 Display Question Number : Yes Is Question Mandatory : No

The primary mineral of zinc is:

**Options :**

1. ✘ Monazite
2. ✔ Sphalerite
3. ✘ Rhodonite
4. ✘ Brucite

**Question Number : 105 Question Id : 47720321337 Display Question Number : Yes Is Question Mandatory : No**

For processing of lean ores of commercially pure metals, which of the following route is ideally suitable:

**Options :**

1. ✘ Pyrometallurgy
2. ✔ Hydrometallurgy
3. ✘ Electrometallurgy
4. ✘ Hydrometallurgy and Electrometallurgy

**Question Number : 106 Question Id : 47720321338 Display Question Number : Yes Is Question Mandatory : No**

Which of the following elements is not primarily a heat producing element in a fuel ?

**Options :**

1. ✘ Carbon



2. ✘ Silicon

3. ✔ Iron

4. ✘ Phosphorous

**Question Number : 107 Question Id : 47720321339 Display Question Number : Yes Is Question Mandatory : No**

The temperature to analyse VCM in coal should be:

**Options :**

1. ✘ 110 °C

2. ✔ 950 °C

3. ✘ 400 °C

4. ✘ 800 °C

**Question Number : 108 Question Id : 47720321340 Display Question Number : Yes Is Question Mandatory : No**

The maximum tolerable limit of sulfur in a good metallurgical coke is:

**Options :**

1. ✘ 4 %

2. ✘ < 3%

3. ✓ 0.5 – 3%

4. ✘ 2.1 %

**Question Number : 109 Question Id : 47720321341 Display Question Number : Yes Is Question Mandatory : No**

The fuel with highest calorific value is:

**Options :**

1. ✘ Briquette

2. ✓ Natural gas

3. ✘ Petroleum

4. ✘ Wood

**Question Number : 110 Question Id : 47720321342 Display Question Number : Yes Is Question Mandatory : No**

A neutral refractory:

**Options :**

1. ✘ Dinas brick

2. ✘ Alumina bricks

3. ✓ Chromite bricks

4. ✘ Magnesia bricks

**Question Number : 111 Question Id : 47720321343 Display Question Number : Yes Is Question Mandatory : No**

An ideal instrument to measure temperature above 1200 °C in a Metallurgical process:

**Options :**

1. ✘ Mercury thermometer
2. ✔ Thermoelectric pyrometer
3. ✘ Gas thermometer
4. ✘ Resistance thermometer

**Question Number : 112 Question Id : 47720321344 Display Question Number : Yes Is Question Mandatory : No**

A system which can not exchange matter but energy with its surroundings is:

**Options :**

1. ✘ Isolated system
2. ✔ Closed system
3. ✘ Open system
4. ✘ Isobaric system

Question Number : 113 Question Id : 47720321345 Display Question Number : Yes Is Question

Mandatory : No

$\Delta H_{298}^0 = -400 \text{ kCal}$  for alumina corresponds to:

Options :

1. ✓ Standard heat of formation
2. ✗ Heat of conservation
3. ✗ Reaction heat generated
4. ✗ Entropy of formation

Question Number : 114 Question Id : 47720321346 Display Question Number : Yes Is Question

Mandatory : No

At equilibrium the entropy of a closed system is:

Options :

1. ✓ Maximum
2. ✗ Significant
3. ✗ Minimum
4. ✗ Constant

Question Number : 115 Question Id : 47720321347 Display Question Number : Yes Is Question

Mandatory : No

The specific heat capacity of water at 25 °C is:

Options :

1. ✓ 4184 J.kg<sup>-1</sup>K<sup>-1</sup>
2. ✗ 4814 J.kg<sup>-1</sup>K<sup>-1</sup>
3. ✗ 418.4 J.kg<sup>-1</sup>K<sup>-1</sup>
4. ✗ 481.4 J.kg<sup>-1</sup>K<sup>-1</sup>

Question Number : 116 Question Id : 47720321348 Display Question Number : Yes Is Question Mandatory : No

Which of the following statement is true:

Options :

1. ✓ Gibbs free energy does not have absolute value
2. ✗ In a vapor phase, fugacity is equal to partial pressure
3. ✗ For an ideal Raoultian behavior activity co-efficient is greater than 1
4. ✗ For an ideal Henerian behavior activity co-efficient is greater than 1

Question Number : 117 Question Id : 47720321349 Display Question Number : Yes Is Question Mandatory : No

The degree of freedom at triple point in a phase diagram:

Options :

1. ✗ 3

2. ✘ 2

3. ✘ 1

4. ✔ 0

**Question Number : 118 Question Id : 47720321350 Display Question Number : Yes Is Question Mandatory : No**

In Ellingham diagram of oxides, the downward slope CO signifies

**Options :**

1. ✘ Decrease in entropy
2. ✘ Can reduce all the oxides at all temperatures
3. ✘ Effective reducing agent above  $717^{\circ}\text{C}$
4. ✔ The most efficient reducing agent for oxides

**Question Number : 119 Question Id : 47720321351 Display Question Number : Yes Is Question Mandatory : No**

Enthalpy is expressed as:

**Options :**

1. ✔  $H = E - PV$
2. ✘  $H - E = PV$

3. ✘  $H = F - TS$

4. ✘  $H + F = TS$

Question Number : 120 Question Id : 47720321352 Display Question Number : Yes Is Question Mandatory : No

The atomic diameter of a FCC crystal with lattice parameter  $a$  is:

Options :

1. ✔  $a\sqrt{2}/2$

2. ✘  $a\sqrt{2}/4$

3. ✘  $a\sqrt{3}/4$

4. ✘  $a/2$

Question Number : 121 Question Id : 47720321353 Display Question Number : Yes Is Question Mandatory : No

The number of atoms along the body diagonal of the diamond cubic unit cell is

Options :

1. ✘ 1

2. ✔ 2

3. ✘ 3

4. ✘ 4

**Question Number : 122 Question Id : 47720321354 Display Question Number : Yes Is Question Mandatory : No**

In deciding the solid solubility (Hume-Ruthery rule), the difference between the atomic diameter of the solute and solvent should not be more than

**Options :**

1. ✘ 50%

2. ✔ 15%

3. ✘ 2%

4. ✘ 0%

**Question Number : 123 Question Id : 47720321355 Display Question Number : Yes Is Question Mandatory : No**

Relative amount of phases in a region in a phase diagram can be estimated by:

**Options :**

1. ✘ Phase rule

2. ✘ Tie-line rule

3. ✘ Humerothery rule

4. ✔ Lever rule



**Question Number : 124 Question Id : 47720321356 Display Question Number : Yes Is Question Mandatory : No**

The reaction of generation of one solid and liquid phase from a solid phase on heating is known as:

**Options :**

1. ✘ Eutectic
2. ✘ Eutectoid
3. ✘ Peritectoid
4. ✔ Peritectic

**Question Number : 125 Question Id : 47720321357 Display Question Number : Yes Is Question Mandatory : No**

The fraction of pearlite in a 0.55% C steel is:

**Options :**

1. ✘ 0.55
2. ✘ 0.31
3. ✘ 0
4. ✔ 0.69

**Question Number : 126 Question Id : 47720321358 Display Question Number : Yes Is Question**

**Mandatory : No**

The unit of flux  $J$  is:

**Options :**

1. ✓ atoms  $\text{m}^{-2} \text{s}^{-1}$

2. ✗ atoms  $\text{m}^2 \text{s}^{-1}$

3. ✗ moles  $\text{m}^2 \text{s}^{-1}$

4. ✗ moles  $\text{m}^{-3} \text{s}^{-1}$

**Question Number : 127 Question Id : 47720321359 Display Question Number : Yes Is Question**

**Mandatory : No**

Which of the following elements has the highest diffusion coefficient in steel at 1000 °C?

**Options :**

1. ✗ Mn

2. ✗ W

3. ✗ Ni

4. ✓ C

**Question Number : 128 Question Id : 47720321360 Display Question Number : Yes Is Question**

**Mandatory : No**

The condition for fine grain size during solidification would be:

**Options :**

1. ✘ Slow cooling
2. ✘ Increasing surface energy
3. ✘ Decreasing nucleation rate
4. ✔ Fast cooling

Question Number : 129 Question Id : 47720321361 Display Question Number : Yes Is Question Mandatory : No

Bainite has:

Options :

1. ✘ Same morphology as austenite
2. ✔ A non-lamellar morphology of ferrite and cementite
3. ✘ The coarsest morphology in the Fe-C diagram
4. ✘ The hardest phase

Question Number : 130 Question Id : 47720321362 Display Question Number : Yes Is Question Mandatory : No

Materials with metallic bonds in its atoms are necessarily

Options :

1. ✔ Ductile under stress
2. ✘ Hard

3. ✘ Gases at RT

4. ✘ Low in electrical conductivity

**Question Number : 131 Question Id : 47720321363 Display Question Number : Yes Is Question Mandatory : No**

Which of the following phase is obtained as the end product in steel, after completion of austempering process?

**Options :**

1. ✘ Austenite

2. ✔ Bainite

3. ✘ Martensite

4. ✘ Pearlite

**Question Number : 132 Question Id : 47720321364 Display Question Number : Yes Is Question Mandatory : No**

Identify the wrong statement pertaining to heat treatment of steel.

**Options :**

1. ✘ Martempering process is designed to overcome limitations of quenching

2. ✔ Pearlite is obtained as the final phase in martempering process

3. ✘ Water is used as quenching medium in Jominy end quench test

4. ✘ Martensite is the end product in steel after austempering

Question Number : 133 Question Id : 47720321365 Display Question Number : Yes Is Question

Mandatory : No

TTT diagram is also known as:

Options :

1. ✔ Bain's curve

2. ✘ S-N curve

3. ✘ Evans curve

4. ✘ Kellog's diagram

Question Number : 134 Question Id : 47720321366 Display Question Number : Yes Is Question

Mandatory : No

% C in medium carbon steels range from:

Options :

1. ✘ 0.1 – 0.2

2. ✘ 0.2 – 0.25

3. ✔ 0.3 – 0.6

4. ✘ 0.7 – 0.8

**Question Number : 135 Question Id : 47720321367 Display Question Number : Yes Is Question Mandatory : No**

A given component cracked after heat treatment. What can be the possible reason?

**Options :**

1. ✘ Prolonged heating
2. ✘ Slow cooling in air
3. ✘ Improper cleaning
4. ✔ Sudden cooling in brine solution

**Question Number : 136 Question Id : 47720321368 Display Question Number : Yes Is Question Mandatory : No**

The austenitizing temperature (for full annealing) for hypo-eutectoid steel is in the range of:

**Options :**

1. ✔ 723 – 910 °C
2. ✘ 910 – 1130 °C
3. ✘ 467 – 723 °C
4. ✘ 668 – 800 °C

**Question Number : 137 Question Id : 47720321369 Display Question Number : Yes Is Question**

**Mandatory : No**

Quench hardening of a steel would produce a hardness depending upon:

**Options :**

1. ✘ Rate of heating
2. ✔ Quenching temperature
3. ✘ Quenching pressure
4. ✘ Water

**Question Number : 138 Question Id : 47720321370 Display Question Number : Yes Is Question**

**Mandatory : No**

Which type of stainless steel has the highest corrosion resistance?

**Options :**

1. ✘ Martensite
2. ✘ Ferrite
3. ✔ Austenite
4. ✘ Dual phase steel

**Question Number : 139 Question Id : 47720321371 Display Question Number : Yes Is Question**

**Mandatory : No**

Manganese addition to steel:

**Options :**

1. ✘ Promotes grain coarsening
2. ✔ Counters effect of sulphur
3. ✘ Increases corrosion resistance
4. ✘ Increases ductility

Question Number : 140 Question Id : 47720321372 Display Question Number : Yes Is Question Mandatory : No

The medium used in pack carburising of steel:

Options :

1. ✔ Activated charcoal
2. ✘ Hydrocarbon gas
3. ✘ Fused salt
4. ✘ Mixture of gas and charcoal

Question Number : 141 Question Id : 47720321373 Display Question Number : Yes Is Question Mandatory : No

The season cracking in yellow  $\alpha$  brasses can be avoided by:

Options :

1. ✘ Full annealing

2. ✘



## Tempering

3. ✓ Stress relief annealing
4. ✗ Age hardening

**Question Number : 142 Question Id : 47720321374 Display Question Number : Yes Is Question Mandatory : No**

Quenching medium with the least severity effect:

**Options :**

1. ✗ Brine
2. ✗ Soluble oil
3. ✗ Liquid salts
4. ✓ Air

**Question Number : 143 Question Id : 47720321375 Display Question Number : Yes Is Question Mandatory : No**

The majority and widest iron bearing mineral is:

**Options :**

1. ✓ Hematite
2. ✗ Limonite

3. ✘ Magnetite

4. ✘ Siderite

**Question Number : 144 Question Id : 47720321376 Display Question Number : Yes Is Question Mandatory : No**

The main role of flux addition during BF iron making is:

**Options :**

1. ✘ To increase the softening point of gangue

2. ✔ To increase the chemical potential of impurities in pure metal

3. ✘ To increase the viscosity of slag

4. ✘ Alloy formation

**Question Number : 145 Question Id : 47720321377 Display Question Number : Yes Is Question Mandatory : No**

An example of dry BF gas cleaning equipment

**Options :**

1. ✘ Venturi scrubber

2. ✘ Scrubber

3. ✘ Hydrocyclone

4. ✓ Dust catcher

**Question Number : 146 Question Id : 47720321378 Display Question Number : Yes Is Question Mandatory : No**

The deciding factor of Mn content in pig iron:

**Options :**

1. ✗ Slag viscosity
2. ✗ CaO content of the burden
3. ✗ Operating pressure temperature
4. ✓ Slag basicity

**Question Number : 147 Question Id : 47720321379 Display Question Number : Yes Is Question Mandatory : No**

Which is closest to the pure form of iron?

**Options :**

1. ✗ Cast iron
2. ✗ Pig iron
3. ✓ Wrought iron
4. ✗ Steel

Question Number : 148 Question Id : 47720321380 Display Question Number : Yes Is Question

Mandatory : No

The product of a commercial direct reduction process is:

Options :

1. ✘ Liquid iron

2. ✘ Iron saturated with carbon

3. ✘ Pig iron

4. ✔ Sponge iron

Question Number : 149 Question Id : 47720321381 Display Question Number : Yes Is Question

Mandatory : No

Which of the following is not an irregularity in a BF operation?

Options :

1. ✘ Hanging

2. ✘ Breakout

3. ✘ Slipping

4. ✔ Tapping

Question Number : 150 Question Id : 47720321382 Display Question Number : Yes Is Question

Mandatory : No

The process which can be used to produce alloy steel:

**Options :**

1. ✘ L D process
2. ✔ Electric arc process
3. ✘ Open hearth process
4. ✘ Acid Bessemer process

**Question Number : 151 Question Id : 47720321383 Display Question Number : Yes Is Question Mandatory : No**

The process for direct smelting to produce iron:

**Options :**

1. ✔ COREX
2. ✘ Vacuum arc degassing
3. ✘ BF process
4. ✘ LD process

**Question Number : 152 Question Id : 47720321384 Display Question Number : Yes Is Question Mandatory : No**

The reactor mechanism during LD steel making process:

**Options :**

1. ✘ Fluidized bed reactor

2. ✘ Retort

3. ✔ Pneumatic reactor

4. ✘ Travelling grate reactor

**Question Number : 153 Question Id : 47720321385 Display Question Number : Yes Is Question Mandatory : No**

The approximate thickness (in mm) of the skin of casting formed at the initial stage continuous casting process is:

**Options :**

1. ✘ 1 – 5

2. ✔ 10 – 25

3. ✘ 75 – 150

4. ✘ 25 – 75

**Question Number : 154 Question Id : 47720321386 Display Question Number : Yes Is Question Mandatory : No**

Reasons for casting defects like diagonal cracks and blows:

**Options :**

1. ✘ Oscillation of mould

2. ✘ High moisture

3. ✓ Mechanical and thermal stress

4. ✗ High heating rate

**Question Number : 155 Question Id : 47720321387 Display Question Number : Yes Is Question Mandatory : No**

India is the third largest global producer of:

**Options :**

1. ✗ Copper

2. ✗ Thorium

3. ✓ Aluminium

4. ✗ Zinc

**Question Number : 156 Question Id : 47720321388 Display Question Number : Yes Is Question Mandatory : No**

The temperature of operation during Pidgeon operation is:

**Options :**

1. ✗ 800 – 700 °C

2. ✗ 900 – 1000 °C

3. ✗ 1000 – 1100 °C

4. ✓ 1100 – 1200 °C

**Question Number : 157 Question Id : 47720321389 Display Question Number : Yes Is Question Mandatory : No**

The bauxite deposits in Kashmir is not suitable for Al production, due to:

**Options :**

1. ✗ Anode effects

2. ✗ High  $\text{TiO}_2$  content

3. ✗ Generation of anode slime

4. ✓ Forms red mud

**Question Number : 158 Question Id : 47720321390 Display Question Number : Yes Is Question Mandatory : No**

The maximum % of dissolution of  $\text{Al}_2\text{O}_3$  in  $3\text{NaF} \cdot \text{AlF}_3$

**Options :**

1. ✗ 5

2. ✓ 15

3. ✗ 0.5

4. ✗ 51



Question Number : 159 Question Id : 47720321391 Display Question Number : Yes Is Question Mandatory : No

The admissible % of Cu in a commercial grade ore is:

Options :

1. ✘ 32 – 35

2. ✔ 0.5 – 2

3. ✘ 70 – 75

4. ✘ 55 – 56

Question Number : 160 Question Id : 47720321392 Display Question Number : Yes Is Question Mandatory : No

Identify the correct statement:

Options :

1. ✔ The purpose of roasting Cu sulfide ore is to partially oxidize iron sulfide present in the ore

2. ✘ Cu smelting process takes place in a blast furnace

3. ✘ For high grade copper sulfide ore roasting is also required

4. ✘ The byproduct of Cu extraction process is lead



Question Number : 161 Question Id : 47720321393 Display Question Number : Yes Is Question Mandatory : No

The composition of Fayalite is:

Options :

1. ✘ Feo.SiO<sub>2</sub>
2. ✔ 2Feo.SiO<sub>2</sub>
3. ✘ 2Feo.2SiO<sub>2</sub>
4. ✘ Feo.2SiO<sub>2</sub>

Question Number : 162 Question Id : 47720321394 Display Question Number : Yes Is Question Mandatory : No

Imperial smelting BF is used for pyro-metallurgical extraction of:

Options :

1. ✔ Zinc
2. ✘ Cu
3. ✘ Al
4. ✘ Th

Question Number : 163 Question Id : 47720321395 Display Question Number : Yes Is Question Mandatory : No

$TiCl_4(l) + 2Mg(l) \xrightarrow{800^\circ C} Ti(C) + 2MgCl_2(l)$  is known as

Options :

1. ✘ Hunter's process

2. ✘ Sorel Process

3. ✔ Kroll's process

4. ✘ Bayer's process

**Question Number : 164 Question Id : 47720321396 Display Question Number : Yes Is Question Mandatory : No**

Monazite is a mineral of :

**Options :**

1. ✘ Uranium only

2. ✘ Titanium only

3. ✔ Uranium and Thorium

4. ✘ Magnesium

**Question Number : 165 Question Id : 47720321397 Display Question Number : Yes Is Question Mandatory : No**

Which of the following is true for creep?

**Options :**

1. ✔ The slope of the strain-time graph increases with temperature and stress

2. ✘ The slope of strain-time graph decreases with stress

3. ✘ The slope of strain-time graph decreases with temperature

4. ✘ The slope of strain-time graph does not depend on temperature or stress

**Question Number : 166 Question Id : 47720321398 Display Question Number : Yes Is Question Mandatory : No**

Which of the following is responsible for fatigue failure?

**Options :**

1. ✘ A minimum tensile stress of sufficiently high value

2. ✔ A sufficiently large number of cycles of applied stress

3. ✘ An uniform variation in applied stress

4. ✘ No compression component

**Question Number : 167 Question Id : 47720321399 Display Question Number : Yes Is Question Mandatory : No**

Which of the hardness tester is the best to obtain bulk hardness of a particulate reinforced metal matrix composite?

**Options :**

1. ✘ Rockwell hardness tester

2. ✓ Brinell hardness tester
3. ✗ Vickers micro hardness tester
4. ✗ Micro-hardness tester

**Question Number : 168 Question Id : 47720321400 Display Question Number : Yes Is Question Mandatory : No**

Which one of the following cannot be determined from tensile test data?

**Options :**

1. ✗ Stiffness
2. ✗ Ductility
3. ✗ Toughness
4. ✓ Malleability

**Question Number : 169 Question Id : 47720321401 Display Question Number : Yes Is Question Mandatory : No**

In Izod test, the specimen is kept as

**Options :**

1. ✗ Simply supported beam
2. ✗ Overhanging beam
3. ✓

Cantilever beam

4. ✘ Fixed ended beam

**Question Number : 170 Question Id : 47720321402 Display Question Number : Yes Is Question Mandatory : No**

What principle defines eddy current inspection (ECI)

**Options :**

1. ✘ Lenz law

2. ✘ Faraday's law

3. ✘ Biot-Savart law

4. ✔ Electromagnetic induction principle

**Question Number : 171 Question Id : 47720321403 Display Question Number : Yes Is Question Mandatory : No**

In radiography test, which of the following samples can be tested?

**Options :**

1. ✘ Metal billets

2. ✘ Metallic foams

3. ✔ Metal sheets

4. ✘

## Metal Matrix Composites

**Question Number : 172 Question Id : 47720321404 Display Question Number : Yes Is Question Mandatory : No**

What is the drawback of ultrasonic testing?

**Options :**

1. ✘ Low depth
2. ✘ Shape restriction
3. ✘ Higher errors
4. ✔ High sensitivity

**Question Number : 173 Question Id : 47720321405 Display Question Number : Yes Is Question Mandatory : No**

Which one is not an example of interstitial impurity?

**Options :**

1. ✘ N in Fe
2. ✘ H in Pd
3. ✔ Cu in Al
4. ✘ C in Fe

Question Number : 174 Question Id : 47720321406 Display Question Number : Yes Is Question

Mandatory : No

What term is used for the defect produced by array of dislocations that produces a small difference in orientation between the adjoining lattice?

Options :

1. ✘ Free surface
2. ✘ Twist boundary
3. ✘ Tilt boundary
4. ✔ Low angle grain boundary

Question Number : 175 Question Id : 47720321407 Display Question Number : Yes Is Question

Mandatory : No

During cold deformation, work hardening occurs because of

Options :

1. ✘ Slip plane decreases
2. ✔ Dislocation interaction
3. ✘ Dislocation solute interaction
4. ✘ Dislocation movement

Question Number : 176 Question Id : 47720321408 Display Question Number : Yes Is Question

Mandatory : No



Which one of the following does not introduce imperfection in metals?

**Options :**

1. ✘ Deformation
2. ✔ Annealing
3. ✘ Quenching
4. ✘ Alloying

**Question Number : 177 Question Id : 47720321409 Display Question Number : Yes Is Question Mandatory : No**

Which one of the following defects is not beneficial in thermo-mechanical treatment?

**Options :**

1. ✘ Segregation
2. ✘ Vacancy
3. ✘ Edge dislocation
4. ✔ Porosity

**Question Number : 178 Question Id : 47720321410 Display Question Number : Yes Is Question Mandatory : No**

A Burgers vector represents the extent of

**Options :**

1. ✔

Slip

2. ✘ Elastic deformation

3. ✘ Hardness

4. ✘ Twinning

**Question Number : 179 Question Id : 47720321411 Display Question Number : Yes Is Question Mandatory : No**

Alligating defect occurs during

**Options :**

1. ✘ Extrusion of hot billet

2. ✘ Wire drawing of soft rods

3. ✔ Rolling of unhomogenized slab

4. ✘ Forging of dissimilar metals

**Question Number : 180 Question Id : 47720321412 Display Question Number : Yes Is Question Mandatory : No**

Formation of metal powder to use in powder metallurgy by reducing some compound with CO or other molecules is known as?

**Options :**

1. ✘ Atomization

2. ✘ Crushing
3. ✔ Reduction
4. ✘ Electrolysis

**Question Number : 181 Question Id : 47720321413 Display Question Number : Yes Is Question Mandatory : No**

Sintering is done to \_\_\_\_\_

**Options :**

1. ✔ Increase final strength
2. ✘ initially increase and then to decrease the strength
3. ✘ Decrease final strength
4. ✘ initially decrease and then to increase the strength

**Question Number : 182 Question Id : 47720321414 Display Question Number : Yes Is Question Mandatory : No**

The process of infiltration in sintered products is to improve

**Options :**

1. ✔ Porosity
2. ✘ Surface finish
3. ✘

Dimensional accuracy

4. ✘ Coherent property

**Question Number : 183 Question Id : 47720321415 Display Question Number : Yes Is Question Mandatory : No**

Which of the following pattern allowances are dependent of each other?

**Options :**

1. ✘ Shrinkage and Machine finish allowance

2. ✘ Distortion and shaking allowance

3. ✘ Pattern allowance and shrinkage draft

4. ✔ Shaking allowance and pattern draft

**Question Number : 184 Question Id : 47720321416 Display Question Number : Yes Is Question Mandatory : No**

During heat treatment of aircraft component, swelling was observed. Which of the following cast defect is responsible for such behavior?

**Options :**

1. ✔ Blowholes

2. ✘ Shrinkage

3. ✘ Cracks

4. ✘ Segregation

**Question Number : 185 Question Id : 47720321417 Display Question Number : Yes Is Question Mandatory : No**

Which of the following metals cannot be casted by ceramic mould casting?

**Options :**

1. ✘ Aluminium

2. ✘ Tin

3. ✔ Magnesium

4. ✘ Steel

**Question Number : 186 Question Id : 47720321418 Display Question Number : Yes Is Question Mandatory : No**

A solid aluminium disc of one-meter diameter has to be casted. According to you which one of the following casting processes is most suitable?

**Options :**

1. ✘ Centrifuging casting

2. ✔ Semi Centrifugal Casting

3. ✘ True Centrifugal Casting

4. ✘ High pressure Die Casting

**Question Number : 187 Question Id : 47720321419 Display Question Number : Yes Is Question Mandatory : No**

Which one of the following moulding machines can be used to form mould of very complex shaped pattern?

**Options :**

1. ✘ Jolt Machine
2. ✘ Contoured squeeze
3. ✔ Diaphragm squeeze
4. ✘ Slings

**Question Number : 188 Question Id : 47720321420 Display Question Number : Yes Is Question Mandatory : No**

Which of the following moulding properties is essential to eliminate blowholes?

**Options :**

1. ✘ Hot strength
2. ✔ Permeability
3. ✘ Surface finish
4. ✘ Degassing



**Question Number : 189 Question Id : 47720321421 Display Question Number : Yes Is Question**

**Mandatory : No**

Collapsibility is required to break

**Options :**

1. ✘ The pattern to create mould cavity
2. ✘ The mould to take the pattern out
3. ✔ The mould to take the casting out
4. ✘ The mould and pattern

**Question Number : 190 Question Id : 47720321422 Display Question Number : Yes Is Question**

**Mandatory : No**

Which one of the following works as riser?

**Options :**

1. ✔ Hot top
2. ✘ Sleeves
3. ✘ Ingates
4. ✘ Runners

**Question Number : 191 Question Id : 47720321423 Display Question Number : Yes Is Question**

**Mandatory : No**

Chock is used in gating system to

**Options :**

1. ✓ Control pressure of the melt
2. ✘ Distribute melt to mould cavity
3. ✘ Remove impurities in the melt
4. ✘ Pour melt into sprue

Question Number : 192 Question Id : 47720321424 Display Question Number : Yes Is Question Mandatory : No

In welding arc, heat generation is

Options :

1. ✘ Equal everywhere
2. ✘ At cathode
3. ✓ Maximum at anode
4. ✘ At mid-arc

Question Number : 193 Question Id : 47720321425 Display Question Number : Yes Is Question Mandatory : No

Which is not a welding defect?

Options :

1. ✘ Under act



2. ✘ Overlap

3. ✘ Spatter

4. ✔ Precipitation

**Question Number : 194 Question Id : 47720321426 Display Question Number : Yes Is Question Mandatory : No**

Which is not a solid state welding process?

**Options :**

1. ✘ Ultrasonic welding

2. ✔ Electron beam welding

3. ✘ Explosive welding

4. ✘ Friction welding

**Question Number : 195 Question Id : 47720321427 Display Question Number : Yes Is Question Mandatory : No**

Main advantage of gas welding process is that it

**Options :**

1. ✘ Provide high rate of heat input

2. ✔ Is cheap

3. ✘ Gives very strong joint in thicker materials

4. ✘ Provides narrow HAZ

**Question Number : 196 Question Id : 47720321428 Display Question Number : Yes Is Question Mandatory : No**

Which one is the most weldable among the following metals ?

**Options :**

1. ✘ Tool steel

2. ✔ Low carbon steel

3. ✘ Stainless steel

4. ✘ Aluminium

**Question Number : 197 Question Id : 47720321429 Display Question Number : Yes Is Question Mandatory : No**

What type of electrode is not used in TIG welding?

**Options :**

1. ✔ Al-W alloy

2. ✘ Thoriated W

3. ✘ W

4. ✘ Ceriated W

**Question Number : 198 Question Id : 47720321430 Display Question Number : Yes Is Question Mandatory : No**

Oxy-acetylene welding mostly employs \_\_\_\_\_ flame

**Options :**

1. ✘ Oxidizing

2. ✘ Reducing

3. ✔ Neutral

4. ✘ Carburizing

**Question Number : 199 Question Id : 47720321431 Display Question Number : Yes Is Question Mandatory : No**

Among the following welding techniques which is mostly used in automatic set up

**Options :**

1. ✘ Gas welding

2. ✘ TIG

3. ✘ Thermit

4. ✔ MIG

Question Number : 200 Question Id : 47720321432 Display Question Number : Yes Is Question Mandatory : No

Thermit welding is a form of \_\_\_\_\_ welding

Options :

1. ✘ Arc

2. ✔ Thermochemical

3. ✘ Gas

4. ✘ Resistance